

Equilibrium Constant Post Lab Answers

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Lab 1: Chemical Equilibrium: Finding a Constant, K_c In this laboratory experiment, a combination of solution chemistry, stoichiometry and spectrophotometric analysis will be used to determine the equilibrium constant for a reaction between iron (III) ion (Fe^{3+}) and thiocyanate ion (SCN^-). In acidic solution, these ions form a blood-red complex ion as shown in equation 4: (4) Lab 5 - Determination of an Equilibrium Constant Question: Laboratory 6 Spectroscopic Determination Of An Equilibrium Constant DATE: NAME SECTION: INSTRUCTOR In-Lab Guidelines (This Is A Guide To The Information That You Should Be Observing While In The Laboratory. Record This Information In The Laboratory Notebook. Turn In The Duplicate Period.) E Pages Of Your Observations To Your Instructor Before You Leave ... Solved:

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